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#### Abstract

Triazapentadienium iodides $\mathbf{2}$ prepared from $N^{\prime}$-thiocarbamoylformamidines $\mathbf{1}$ are efficient intermediates in heterocyclic synthesis. They react with ketenes, sulfenes, phenyl isocyanate or isothiocyanate and dimethyl acetylenedicarboxylate affording the corresponding dihydropyrimidinones $\mathbf{3}$, thiadiazinedioxides 5, triazinones 6, triazinethiones 7 and pyrimidines 9.


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The synthetic value of 1,3-diazadienes in heterocyclic chemistry has been widely demonstrated [1]. During the last fifteen years $[4+2]$ cycloaddition reactions of acyclic 1,3-diazadienes, especially with ketenes, have been studied giving rise to various dihydropyrimidines [2-9].

The problems encountered in the preparation of stable 1 -substituted-1,3-diazadienes were recently solved by the use of cationic precursors [10].


We report here the reactivity of triazadienium iodides 2 in [4 +2] cycloaddition reactions affording methylsulfanyldihydropyrimidinones $\mathbf{3}$ and thioxotetrahydropyrimidinones $\mathbf{4}$, thiadiazinedioxides 5, triazinones 6, triazinethiones $\mathbf{7}$ and pyrimidines 9 .
Scheme 2


| Compound | $\mathrm{R}^{1}$ | $\mathrm{R}^{2}$ | Yield (\%) | $\mathrm{Mp}\left({ }^{\circ} \mathrm{C}\right)$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathbf{3 a}$ | H | $\mathrm{CO}_{2} \mathrm{CH}_{3}$ | 82 | 205 |
| $\mathbf{3 b}[18]$ | H | $\mathrm{CO}_{2} \mathrm{C}_{2} \mathrm{H}_{5}$ | 63 | 133 |
| $\mathbf{3 c}$ | $\mathrm{CH}_{3}$ | $\mathrm{CO}_{2} \mathrm{CH}_{3}$ | 84 | 179 |
| $\mathbf{3 d}$ | $\mathrm{CH}_{3}$ | $\mathrm{CO}_{2} \mathrm{C}_{2} \mathrm{H}_{5}$ | 84 | 149 |
| $\mathbf{3 e}$ | $\mathrm{CH}_{3}$ | $\mathrm{C}_{6} \mathrm{H}_{5}$ | 97 | 196 |
| $\mathbf{4 a}$ | H | $\mathrm{CO}_{2} \mathrm{CH}_{3}$ | 72 | 181 |
| $\mathbf{4 b}[19]$ | H | $\mathrm{CO}_{2} \mathrm{C}_{2} \mathrm{H}_{5}$ | 77 | 244 |
| $\mathbf{4 c}$ | $\mathrm{CH}_{3}$ | $\mathrm{CO}_{2} \mathrm{CH}_{3}$ | 90 | 232 |
| $\mathbf{4 d}$ | $\mathrm{CH}_{3}$ | $\mathrm{CO}_{2} \mathrm{C}_{2} \mathrm{H}_{5}$ | 97 | 211 |
| $\mathbf{4 e}[20]$ | $\mathrm{CH}_{3}$ | $\mathrm{C}_{6} \mathrm{H}_{5}$ | 98 | 289 |

These types of compounds exhibit important pharmacological potential. Recently $S$-alkylsufanylpyrimidine and $S$-alkylsulfanylpyrimidinone derivatives have been identified as highly specific reverse transcriptase inhibitors of human immunodeficiency virus type 1 [11,12]. Thioxo derivatives showed a diverse range of biological properties such as hyperthyroidism treatment, anticarcinogenic, cardiovascular and antihypertensive activities [13-14]. To our knowledge, only few monocyclic thiadiazinedioxides were previously reported $[10,15]$ even though fused pyrido derivatives were found to be important in several pharmacological fields [16].

We have prepared $N^{\prime}$-thiocarbamoylformamidines 1 following Bredereck procedure [17]. Subsequent alkylation using methyliodide led to $S$-methyl salts 2 in high yields (Scheme 1).

Ketenes were prepared in situ from acid chlorides (methyl or ethyl malonyl chloride, phenylacetyl chloride) and reacted with $S$-methyliodides 2 to give the corresponding dihydropyrimidinones $\mathbf{3}$. The $[4+2]$ cycloaddition proceeded smoothly and was followed by loss of dimethylamine. Addition of triethylamine was necessary to neutralize the two equivalents of the hydracids generated in these reactions.

Then the methylsulfanylimine group of compounds $\mathbf{3}$ was eliminated by treatment with hydrogen sulfide to yield thioxotetrahydropyrimidinones 4 (Scheme 2).

Scheme 3


Scheme 4

| $2+\mathrm{C}_{6} \mathrm{H}_{5}-\mathrm{N}=\mathrm{C}=\mathrm{X}$ | $\longrightarrow$ |
| :---: | :---: | :---: | :---: | :---: |

Sulfenes prepared from the corresponding sulfonic acid chlorides (mesyl chloride, ethanesulfonyl chloride, $\alpha$-toluenesulfonyl chloride) reacted readily with salts $\mathbf{2}$, using the same conditions (vide supra), to afford $2 H-1,2,4-$ thiadiazine-1,1-dioxides 5 (Scheme 3).

A cyclization reaction took place with compounds 2 in the presence of phenyl isocyanate or phenyl isothiocyanate resulting in the formation of triazin- $2(1 H)$-ones 6 and triazine-2 1 H )-thiones 7 respectively in good yields (Scheme 4).

Scheme 5


The reaction of salts 2 with dimethyl acetylenedicarboxylate afforded 2-methylsulfanylpyrimidines 9 . Starting from 2b, we have isolated a rather unstable intermediate compound $\mathbf{8}$ identified as a linear adduct between the unsubstituted imine group of the diazadiene and the triple bond of the acetylenic reagent. A ring closure of intermediate $\mathbf{8}$ followed by the loss of dimethylamine provided pyrimidine $\mathbf{9 b}$ on heating at $110^{\circ}$ for 18 hours (Scheme 5).
Although most of our compounds were expected to be generated by a classical Diels-Alder cycloaddition, the later result indicated obviously that in this case the
operative pathway to afford the heterocyclic products is an addition reaction followed by an intramolecular DielsAlder cyclization.

In summary, we have demonstrated the ability of triazapentadienium salts 2 to undergo annulation reactions under rather mild conditions providing an efficient synthetic method for the preparation of various pyrimidine derivatives. Moreover reactions with sulfenes afforded novel $4 H$-1,2,4-thiazine-1,1-dioxides in satisfactory yields.

## EXPERIMENTAL

All reagents were purchased from Acros Organics and Aldrich. Elemental analyses were performed by the C.N.R.S. Analysis Laboratory (Vernaison). Column chromatography was conducted on silica gel $60(40-63 \mu \mathrm{~m})$, available from E . Merck. Thin layer chromatography was performed on 0.5 mm x $20 \mathrm{~cm} \times 20 \mathrm{~cm}$ E. Merck silica gel plate ( 60 F-254). Melting points measured using a Reichert microscope are uncorrected. The ${ }^{13} \mathrm{C}$ and ${ }^{1} \mathrm{H}$-nmr spectra were recorded at room temperature using a BRUKER AC200 at 50 and 200 MHz respectively. Chemical shifts ( $\delta$ ) are given in ppm downfield from tetramethylsilane as internal standard. Mass spectra were determined with a Hewlett Packard 5989 spectrometer. The ir spectra were obtained using a BRUKER Vector22 spectrometer.

General Procedure for the Reaction of Thiourea with Amide Dimethyl Acetals.
$\mathrm{N}, \mathrm{N}$-Dimethyl formamide dimethyl acetal (for 1a, 13 mmoles ) or $N, N$-dimethyl acetamide dimethyl acetal (for $\mathbf{1 b}, 13$ mmoles) was added to a suspension of thiourea ( 10 mmoles ) in methanol (for $\mathbf{1 a}, 10 \mathrm{ml}$ ) or methylene chloride (for $\mathbf{1 b}, 10 \mathrm{ml}$ ). The mixture was heated under reflux for 4 hours. The solvent was removed and the residue was crystallized from methanol (for 1a) or diethyl ether (for $\mathbf{1 b}$ ).

2-Amino-4-dimethylamino-1,3-thiazabuta-1,3-diene (1a).
This compound was obtained in $70 \%$ yield, $\mathrm{mp} 165^{\circ}$ [17].
2-Amino-4-dimethylamino-1,3-thiazapenta-1,3-diene (1b).
This compound was obtained in $76 \%$ yield, mp 128-130 ; ir (potassium bromide): $v \max 3278,3123,1583,1366 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ nmr (hexadeuteriodimethylsulfoxide): $\delta 2.17$ (s, $3 \mathrm{H}, \mathrm{CH}_{3}$ ), 2.93 $\left(\mathrm{s}, 6 \mathrm{H}, \mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right), 7.66$ and $8.05(2 \mathrm{~s}, 2 \mathrm{H}, 2 \mathrm{NH}) ;{ }^{13} \mathrm{C} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta 16.5\left(\mathrm{CH}_{3}\right), 37.9\left(\mathrm{~N}_{\left.\left(\mathrm{CH}_{3}\right)_{2}\right)}\right.$, $159.0\left(\mathrm{CCH}_{3}\right), 194.0(\mathrm{CS}) ; \mathrm{ms}: \mathrm{m} / \mathrm{z} 145\left(\mathrm{M}^{+}\right), 129\left(\mathrm{M}^{+}-\mathrm{NH}_{2}\right)$, $112\left(\mathrm{M}^{+}-\mathrm{SH}\right)$.

Anal. Calcd. for $\mathrm{C}_{5} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{~S}: \mathrm{C}, 41.35 ; \mathrm{H}, 7.63 ; \mathrm{N}, 28.94$. Found: C, 41.05; H, 7.72; N, 29.16.

General Procedure for Addition of Methyl Iodide on Thiazadienes 1.

A suspension of thiazadiene $\mathbf{1}$ ( 5 mmoles ) in methyl iodide $(4 \mathrm{ml})$ and tetrahydrofuran $(4 \mathrm{ml})$ was stirred for 18 hours at room temperature. The mixture was evaporated under reduced pressure. After addition of diethyl ether ( 40 ml ), compounds 2 were precipitated and collected by filtration.

## 1,1-Dimethyl-4-methylsulfanyl-1,3,5-triazapentadienium Iodide (2a).

This compound was prepared from 1a in $98 \%$ yield, $\mathrm{mp} 148^{\circ}$; ir (potassium bromide): $v \max 3232,3070,1624,1521 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ nmr (hexadeuteriodimethylsulfoxide): $\delta 2.55\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{SCH}_{3}\right), 3.11$ and $3.27\left(2 \mathrm{~s}, 6 \mathrm{H}, \mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right), 8.39(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}), 9.53\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{NH}_{2}\right)$; ${ }^{13} \mathrm{C} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta 14.4\left(\mathrm{SCH}_{3}\right), 35.9$ and $41.9\left(\mathrm{~N}\left(\mathrm{CH}_{3}\right)_{2}\right), 158.7(\mathrm{CH}), 178.2(\mathrm{CS})$.

Anal. Calcd. for $\mathrm{C}_{5} \mathrm{H}_{12} \mathrm{IN}_{3} \mathrm{~S}: \mathrm{C}, 21.99 ; \mathrm{H}, 4.43 ; \mathrm{N}, 15.38$. Found: C, 21.71; H, 4.60; N, 15.55.

1,1,2-Trimethyl-4-methylsulfanyl-1,3,5-triazapentadienium Iodide (2b).

This compound was prepared from $\mathbf{1 b}$ in $95 \%$ yield, $\mathrm{mp} 180^{\circ}$; ir (potassium bromide): $v \max 3258,3108,1629,1569 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta 2.29$ (s, $3 \mathrm{H}, \mathrm{CH}_{3}$ ), 2.41 ( $\mathrm{s}, 3 \mathrm{H}$, $\left.\mathrm{SCH}_{3}\right), 3.13$ and $3.23\left(2 \mathrm{~s}, 6 \mathrm{H}, \mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right), 9.01\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{NH}_{2}\right) ;{ }^{13} \mathrm{C}$ nmr (hexadeuteriodimethylsulfoxide): $\delta 14.3\left(\mathrm{SCH}_{3}\right), 18.5\left(\mathrm{CH}_{3}\right)$, 39.2 and $39.8\left(\mathrm{~N}\left(\mathrm{CH}_{3}\right)_{2}\right), 166.4\left(\mathrm{CCH}_{3}\right), 172.0(\mathrm{CS})$.

Anal. Calcd. for $\mathrm{C}_{6} \mathrm{H}_{14} \mathrm{IN}_{3} \mathrm{~S}: \mathrm{C}, 25.10 ; \mathrm{H}, 4.91 ; \mathrm{N}, 14.63$. Found: C, 24.88; H, 5.04; N, 14.61.

General Procedure for the Reaction Between Triazapentadienum Iodides 2 and Acyl Chlorides or Sulfonyl Chlorides.

To a solution of iodide $\mathbf{2}$ ( 2 mmoles ) in dry methylene chloride ( 10 ml ), acyl chloride (for 3, 2.4 mmoles ) or sulfonyl chloride (for 5, 2.4 mmoles) was added under a nitrogen atmosphere. The mixture was stirred for 4 hours at room temperature, cooled to $0^{\circ}$ then triethylamine ( 4.8 mmoles) was added. The reaction mixture was stirred overnight. The solvent was removed and the residue was purified by flash chromatography (silica gel, methylene chloride/ethyl acetate 9/1). Products $\mathbf{3}$ and 5 were crystallized from appropriate solvents.
5-Methoxycarbonyl-2-methylsulfanylpyrimidin-4(3H)-one (3a).
This compound was prepared from 2a and methyl malonyl chloride in $82 \%$ yield, $\mathrm{mp} 205^{\circ}$ (from diethyl ether); ir (potassium bromide): $v \max 1741,1651 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 2.60\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{SCH}_{3}\right), 3.97\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OCH}_{3}\right), 8.75$ (s, $1 \mathrm{H}, \mathrm{CH}$ ), ( NH , exchange); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (deuteriochloroform): $\delta$ $13.0\left(\mathrm{SCH}_{3}\right), 51.6\left(\mathrm{OCH}_{3}\right), 111.5(\mathrm{CCO}), 158.0(\mathrm{NCH}), 159.0$, 164.1 and $168.1(\mathrm{SCN}$ and 2 CO$) ; \mathrm{ms}: \mathrm{m} / \mathrm{z} 200\left(\mathrm{M}^{+}\right), 168$ $\left(\mathrm{M}^{+}-\mathrm{CH}_{3} \mathrm{OH}\right), 112$.
Anal. Calcd. for $\mathrm{C}_{7} \mathrm{H}_{8} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}: \mathrm{C}, 41.99 ; \mathrm{H}, 4.03 ; \mathrm{N}, 13.99$. Found: C, 42.23; H, 3.97; N, 14.27.

5-Ethoxycarbonyl-2-methylsulfanylpyrimidin-4(3H)-one (3b).
This compound was prepared from 2a and ethyl malonyl chloride in $63 \%$ yield, $\mathrm{mp} 133^{\circ}$ (from diethyl ether) [18].

5-Methoxycarbonyl-6-methyl-2-methylsulfanylpyrimidin$4(3 \mathrm{H})$-one (3c).

This compound was prepared from $\mathbf{2 b}$ and methyl malonyl chloride in $84 \%$ yield, $\mathrm{mp} 179^{\circ}$ (from diethyl ether); ir (potassium bromide): $v \max 1734,1635 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 2.57$ and $2.58\left(2 \mathrm{~s}, 6 \mathrm{H}, \mathrm{CH}_{3}\right.$ and $\left.\mathrm{SCH}_{3}\right), 3.97$ (s, $3 \mathrm{H}, \mathrm{OCH}_{3}$ ), 12.34 (broad s, $1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (deuteriochloroform): $\delta 13.6\left(\mathrm{SCH}_{3}\right), 24.2\left(\mathrm{CH}_{3}\right), 52.6\left(\mathrm{OCH}_{3}\right), 110.2$ $(C \mathrm{CO}), 164.6,166.8,167.2$ and $167.4\left(\mathrm{CCH}_{3}, \mathrm{SCN}\right.$ and 2 CO$)$; $\mathrm{ms}: \mathrm{m} / \mathrm{z} 214\left(\mathrm{M}^{+}\right), 182\left(\mathrm{M}^{+}-\mathrm{CH}_{3} \mathrm{OH}\right), 126$.

Anal. Calcd. for $\mathrm{C}_{8} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}: \mathrm{C}, 44.85 ; \mathrm{H}, 4.70 ; \mathrm{N}, 13.08$. Found: C, 45.17; H, 4.50; N, 13.35 .

5-Ethoxycarbonyl-6-methyl-2-methylsulfanylpyrimidin-4(3H)one (3d).

This compound was prepared from $\mathbf{2 b}$ and ethyl malonyl chloride in $84 \%$ yield, mp $149^{\circ}$ (from diethyl ether); ir (potassium bromide): $v \max 1734,1635 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta$ $1.42\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.2 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 2.57$ and $2.58\left(2 \mathrm{~s}, 6 \mathrm{H}, \mathrm{SCH}_{3}\right.$ and $\left.\mathrm{CH}_{3}\right), 4.43\left(\mathrm{q}, 2 \mathrm{H}, \mathrm{J}=7.2 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, $\left(\mathrm{NH}\right.$, exchange); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (deuteriochloroform): $\delta 13.6\left(\mathrm{SCH}_{3}\right), 14.2\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 24.2$ $\left(\mathrm{CCH}_{3}\right), 61.9\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 110.0(\mathrm{CCO}), 165.1$ and $167.1\left(\mathrm{CCH}_{3}\right.$, SCN and 2 CO$)$; ms: m/z $228\left(\mathrm{M}^{+}\right), 182\left(\mathrm{M}^{+}-\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{OH}\right), 126$.

Anal. Calcd. for $\mathrm{C}_{9} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}$ : C, 47.36; H, 5.30; $\mathrm{N}, 12.27$. Found: C, 47.51; H, 5.03; N, 12.42.

6-Methyl-2-methylsulfanyl-5-phenylpyrimidin-4(3H)-one (3e).
This compound was prepared from 2b and phenylacetyl chloride in $97 \%$ yield, mp $196^{\circ}$ (from diethyl ether); ir (potassium bromide): $v$ max $1646 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 2.20\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 2.55\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{SCH}_{3}\right), 7.25-7.47$ (m, 5H, $\mathrm{C}_{6} \mathrm{H}_{5}$ ), 10.27 (broad s, $1 \mathrm{H}, \mathrm{NH}$ ), ${ }^{13} \mathrm{C} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta 12.7\left(\mathrm{SCH}_{3}\right), 22.4\left(\mathrm{CCH}_{3}\right), 120.2$ $\left(\mathrm{CC}_{6} \mathrm{H}_{5}\right), 127.2,128.0$ and 130.3 (5CHar), 134.3 (Car), 148.0 $\left(\mathrm{CCH}_{3}\right), 159.6$ and $162.2(\mathrm{SCN}$ and CO$)$; ms: m/z $232\left(\mathrm{M}^{+}\right)$.

Anal. Calcd. for $\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{OS}: \mathrm{C}, 62.05 ; \mathrm{H}, 5.21 ; \mathrm{N}, 12.06$. Found: C, 61.96; H, 5.18; N, 12.01.

3-Methylsulfanyl-4H-1,2,4-thiadiazine-1,1-dioxide (5a).
This compound was prepared from 2a and methanesulfonyl chloride in $46 \%$ yield, $\mathrm{mp} 144^{\circ}$ (from methylene chloride); ir (potassium bromide): $v$ max 1646, 1570, 1258, $1119 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ nmr (hexadeuteriodimethylsulfoxide): $\delta 2.42\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{SCH}_{3}\right), 6.26$ (d, $1 \mathrm{H}, \mathrm{J}=8.6 \mathrm{~Hz}, \mathrm{SO}_{2} \mathrm{CH}$ ), $7.01(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=8.6 \mathrm{~Hz}, \mathrm{NCH}), 11.71$ (broad s, $1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta$ $13.8\left(\mathrm{SCH}_{3}\right), 103.5\left(\mathrm{SO}_{2} \mathrm{CH}\right), 132.6(\mathrm{NCH}), 160.9(\mathrm{SCN}) ; \mathrm{ms}:$ $\mathrm{m} / \mathrm{z} 178\left(\mathrm{M}^{+}\right), 114\left(\mathrm{M}^{+}-\mathrm{SO}_{2}\right), 81,74$.

Anal. Calcd. for $\mathrm{C}_{4} \mathrm{H}_{6} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}_{2}$ : C, 26.96; H, 3.39; $\mathrm{N}, 15.72$. Found: C, 30.21; H, 3.51; N, 15.55.

6-Methyl-3-methylsulfanyl-4H-1,2,4-thiadiazine-1,1-dioxide (5b).

This compound was prepared from 2a and ethanesulfonyl chloride in $86 \%$ yield, $\mathrm{mp} 221^{\circ}$ (from diethyl ether); ir (potassium bromide): $v$ max 1660, 1574, 1515, 1279, 1132 $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta 1.96$ (d, 3 H , $\left.\mathrm{J}=1.2 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 2.41\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{SCH}_{3}\right), 6.84(\mathrm{q}, 1 \mathrm{H}, \mathrm{J}=1.2 \mathrm{~Hz}$, CH ), 11.69 (broad s, $1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta 10.9\left(\mathrm{CH}_{3}\right), 13.0\left(\mathrm{SCH}_{3}\right), 112.6$ $\left(\mathrm{CCH}_{3}\right), 128.5(\mathrm{NCH}), 159.5(\mathrm{SCN}) ; \mathrm{ms}: \mathrm{m} / \mathrm{z} 192\left(\mathrm{M}^{+}\right), 128$ $\left(\mathrm{M}^{+}-\mathrm{SO}_{2}\right), 95,54$.

Anal. Calcd. for $\mathrm{C}_{5} \mathrm{H}_{8} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}_{2}$ : C, 31.24; H, 4.19; N, 14.57. Found: C, 31.27; H, 4.40; N, 14.68.

3-Methylsulfanyl-6-phenyl-4H-1,2,4-thiadiazine-1,1-dioxide (5c).

This compound was prepared from 2a and $\alpha$-toluenesulfonyl chloride in $77 \%$ yield, mp $149^{\circ}$ (from diethyl ether); ir (potassium bromide): $v \max 1639,1569,1506,1258,1143,1118$
$\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 2.50\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{SCH}_{3}\right), 6.66$ $(\mathrm{d}, 1 \mathrm{H}, \mathrm{J}=5.8 \mathrm{~Hz}, \mathrm{CH}), 7.33-7.54\left(\mathrm{~m}, 5 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{5}\right), 9.43(\mathrm{~d}, 1 \mathrm{H}$, $\mathrm{J}=5.8 \mathrm{~Hz}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta$ $13.1\left(\mathrm{SCH}_{3}\right), 117.6\left(\mathrm{CC}_{6} \mathrm{H}_{5}\right), 127.6,128.4,128.6,129.3$ and 130.1 (5CHar, Car and NCH), 159.4 (SCN); ms: m/z $254\left(\mathrm{M}^{+}\right)$, $190\left(\mathrm{M}^{+}-\mathrm{SO}_{2}\right), 157,117,90$.
Anal. Calcd. for $\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}_{2}$ : C, 47.23; H, 3.96; $\mathrm{N}, 11.01$. Found: C, 47.02; H, 4.13; N, 11.27.

5-Methyl-3-methylsulfanyl-4H-1,2,4-thiadiazine-1,1-dioxide (5d).

This compound was prepared from $\mathbf{2 b}$ and methanesulfonyl chloride in $69 \%$ yield, $\mathrm{mp} 223-225^{\circ}$ (from diethyl ether/methylene chloride $1 / 1$ ); ir (potassium bromide): $v$ max 1653, 1581, $1120 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta 1.98$ (s, $3 \mathrm{H}, \mathrm{CH}_{3}$ ), $2.43\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{SCH}_{3}\right), 6.18(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}), 11.77$ (broad $\mathrm{s}, 1 \mathrm{H}, \mathrm{NH}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta 13.0$ $\left(\mathrm{SCH}_{3}\right), 18.5\left(\mathrm{CH}_{3}\right), 100.9(\mathrm{CH}), 142.7\left(\mathrm{CCH}_{3}\right), 159.2(\mathrm{SCN})$; $\mathrm{ms}: \mathrm{m} / \mathrm{z} 192\left(\mathrm{M}^{+}\right), 128\left(\mathrm{M}^{+}-\mathrm{SO}_{2}\right), 95,54$.
Anal. Calcd. for $\mathrm{C}_{5} \mathrm{H}_{8} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}_{2}$ : C, 31.24; H, 4.19; $\mathrm{N}, 14.57$. Found: C, 31.18; H, 4.32; N, 14.46.

5,6-Dimethyl-3-methylsulfanyl-4H-1,2,4-thiadiazine-1,1dioxide (5e).

This compound was prepared from $\mathbf{2 b}$ and ethanesulfonyl chloride in $39 \%$ yield, $\mathrm{mp} 206^{\circ}$ (from diethyl ether); ir (potassium bromide): $v$ max 1661, 1579, 1509, $1142 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ nmr (deuteriochloroform): $\delta 2.06$ and $2.11\left(2 \mathrm{~s}, 6 \mathrm{H}, \mathrm{NCCH}_{3}\right.$ and $\mathrm{SCCH}_{3}$ ), $2.49\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{SCH}_{3}\right), 9.14$ (broad s, $\left.1 \mathrm{H}, \mathrm{NH}\right) ;{ }^{13} \mathrm{C} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta 8.6\left(\mathrm{SCCH}_{3}\right), 13.0\left(\mathrm{SCH}_{3}\right)$, $15.5\left(\mathrm{NCCH}_{3}\right), 108.2\left(\mathrm{SCCH}_{3}\right), 137.2\left(\mathrm{NCCH}_{3}\right), 158.9(\mathrm{SCN})$; $\mathrm{ms}: \mathrm{m} / \mathrm{z} 206\left(\mathrm{M}^{+}\right), 142\left(\mathrm{M}^{+}-\mathrm{SO}_{2}\right), 109,68$.
Anal. Calcd. for $\mathrm{C}_{6} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}_{2}$ : C, 34.94; H, 4.89; N, 13.58. Found: C, 34.65; H, 4.71; N, 13.24.

5-Methyl-3-methylsulfanyl-6-phenyl-4H-1,2,4-thiadiazine-1,1dioxide (5f).

This compound was prepared from $\mathbf{2 b}$ and $\alpha$-toluenesulfonyl chloride in $66 \%$ yield, $\mathrm{mp}>300^{\circ}$ (from methylene chloride); ir (potassium bromide): $v \max 1653,1582,1509,1274,1131 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta 1.88\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$, 2.47 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{SCH}_{3}$ ), 7.37-7.42 (m, 5H, $\mathrm{C}_{6} \mathrm{H}_{5}$ ), $11.82($ broad s, 1 H , $\mathrm{NH}) ;{ }^{13} \mathrm{C} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta 13.1\left(\mathrm{SCH}_{3}\right)$, $16.8\left(\mathrm{CH}_{3}\right), 114.6\left(\mathrm{CC}_{6} \mathrm{H}_{5}\right), 128.4,128.6,129.0$ and 131.2 (5CHar and Car), $138.9\left(\mathrm{CCH}_{3}\right), 159.2(\mathrm{SCN}) ; \mathrm{ms}: \mathrm{m} / \mathrm{z} 268$ $(\mathrm{M}+), 204\left(\mathrm{M}^{+}-\mathrm{SO}_{2}\right), 171,131,104$.
Anal. Calcd. for $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}_{2}$ : C, 49.23; H, 4.51; N, 10.44. Found: C, 49.51; H, 4.33; N, 10.70.
General Procedure for the Reaction Between Methylsulfanylpyrimidinones $\mathbf{3}$ and Hydrogen Sulfide.

Hydrogen sulfide was passed for 4 hours through a solution of methylsulfanyl-pyrimidinone 3 ( 2 mmoles ) in triethylamine $(7 \mathrm{ml})$ and pyridine ( 7 ml ). The solvents were removed and the residue was purified by flash chromatography (silica gel, methylene chloride/ethyl acetate $1 / 1$ ). The product was crystallized from an appropriate solvent.

5-Methoxycarbonyl-2-thioxo-1,2,3,4-tetrahydropyrimidin-4-one (4a).

This compound was obtained from 3a in $72 \%$ yield, mp $181^{\circ}$ (from ethyl acetate); ir (potassium bromide): $v$ max 1717, 1616, 1549, 1458, 1216, $1165 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta 3.70\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OCH}_{3}\right), 7.96(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}) ; 10.77$ (broad s, $2 \mathrm{H}, 2 \mathrm{NH}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta$ $51.7\left(\mathrm{OCH}_{3}\right), 106.8(\mathrm{CCO}), 147.6(\mathrm{CH}), 157.1(\mathrm{CCO}), 162.8$ (NCO), 176.5 (CS); ms: m/z $186\left(\mathrm{M}^{+}\right), 170,139,69$.

Anal. Calcd. for $\mathrm{C}_{6} \mathrm{H}_{6} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}$ : C, 38.71 ; $\mathrm{H}, 3.25$; $\mathrm{N}, 15.05$. Found: C, 39.98; H, 3.35; N, 15.28.

5-Ethoxycarbonyl-2-thioxo-1,2,3,4-tetrahydropyrimidin-4-one (4b).

This compound was obtained from $\mathbf{3 b}$ in $77 \%$ yield, $\mathrm{mp} 244^{\circ}$ (from ethyl acetate) [19].

5-Methoxycarbonyl-6-methyl-2-thioxo-1,2,3,4-tetrahydro-pyrimidin-4-one (4c).

This compound was obtained from $\mathbf{3 c}$ in $90 \%$ yield, $\mathrm{mp} 232^{\circ}$ (from diethyl ether/methylene chloride $1 / 1$ ); ir (potassium bromide): $v \max 1726,1628,1559,1164 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta 2.50\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 3.73$ ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{OCH}_{3}$ ), 12.61 (broad $\mathrm{s}, 2 \mathrm{H}, 2 \mathrm{NH}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta 17.2\left(\mathrm{CH}_{3}\right), 52.1\left(\mathrm{OCH}_{3}\right)$, $108.9(\mathrm{CCO}), 154.9\left(\mathrm{CCH}_{3}\right), 157.9(\mathrm{CCO}), 164.4(\mathrm{NCO}), 175.4$ (CS); ms: m/z $200\left(\mathrm{M}^{+}\right), 169\left(\mathrm{M}^{+}-\mathrm{CH}_{3} \mathrm{O}\right), 168$.

Anal. Calcd. for $\mathrm{C}_{7} \mathrm{H}_{8} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}: \mathrm{C}, 41.99$; H, 4.03; N, 13.99. Found: C, 42.37; H, 4.27; N, 14.11.

5-Ethoxycarbonyl-6-methyl-2-thioxo-1,2,3,4-tetrahydropyrim-idin-4-one (4d).

This compound was obtained from 3d in $97 \%$ yield, $\mathrm{mp} 211^{\circ}$ (from 50/50 diethyl ether/methylene chloride); ir (potassium bromide): $v \max 1717,1653,1558,1165 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta 1.23(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.2 \mathrm{~Hz}$, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ), $2.20\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 4.20\left(\mathrm{q}, 2 \mathrm{H}, \mathrm{J}=7.2 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, 12.56 (broad s, $2 \mathrm{H}, 2 \mathrm{NH}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (deuteriochloroform): $\delta 14.0$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 17.1\left(\mathrm{CH}_{3}\right), 60.9\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 109.3(\mathrm{CCO}), 154.3$ $\left(\mathrm{CCH}_{3}\right), 157.9(\mathrm{CCO}), 163.8(\mathrm{NCO}), 175.5(\mathrm{CS}) ; \mathrm{ms:} \mathrm{~m} / \mathrm{z} 214$ $\left(\mathrm{M}^{+}\right), 169\left(\mathrm{M}^{+}-\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{O}\right), 142$.

Anal. Calcd. for $\mathrm{C}_{8} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{~S}$ : C, 44.85 ; H, 4.70; N, 13.08. Found: C, 45.12; H, 4.88; N, 13.33.
6-Methyl-5-phenyl-2-thioxo-1,2,3,4-tetrahydropyrimidin-4-one (4e).

This compound was obtained from $\mathbf{3 e}$ in $98 \%$ yield, $\mathrm{mp} 289^{\circ}$ (from diethyl ether/methylene chloride 1/1) [20].
General Procedure for the Reaction Between Triazapentadienium Iodides 2 and Phenyl Isocyanate or Phenyl Isothiocyanate.

A solution of iodide 2 ( 2 mmoles) and phenyl isocyanate or phenyl isothiocyanate ( 2.2 mmoles ) in dry tetrahydrofuran ( 10 ml ) was stirred at room temperature (for compounds 6) or heated under reflux (for compounds 7) for 6 hours under a nitrogen atmosphere. The reaction mixture was cooled to $0^{\circ}$ and triethylamine ( 4.4 mmoles) was added. The mixture was stirred for 18 hours at room temperature. The solvent was removed and
the residue was purified by flash chromatography (silica gel, methylene chloride/ethyl acetate $9: 1$ ). The product was crystallized from an appropriate solvent.

## 4-Methylsulfanyl-1-phenyl-1,3,5-triazin-2(1H)-one ( $6 \mathbf{a}$ ).

This compound was prepared from 2a and phenyl isocyanate in $55 \%$ yield, $\mathrm{mp} 160^{\circ}$ (from diethyl ether); ir (potassium bromide): $v \max 1730,1612,1457,1233 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 2.58\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{SCH}_{3}\right), 7.35-7.55\left(\mathrm{~m}, 5 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{5}\right)$, 8.07 (s, 1H, CH); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (deuteriochloroform): $\delta 14.4\left(\mathrm{SCH}_{3}\right)$, 126.2, 129.7 and 129.8 (5CHar), 136.4 (Car), 151.2 (CO), 155.9 (CH), $184.4(\mathrm{SCN}) ; \mathrm{ms}: \mathrm{m} / \mathrm{z} 219\left(\mathrm{M}^{+}\right), 204\left(\mathrm{M}^{+}-\mathrm{CH}_{3}\right), 104,77$.

Anal. Calcd. for $\mathrm{C}_{10} \mathrm{H}_{9} \mathrm{~N}_{3} \mathrm{OS}: \mathrm{C}, 54.78 ; \mathrm{H}, 4.14$; $\mathrm{N}, 19.16$. Found: C, 54.58; H, 4.32; N, 18.87.

6-Methyl-4-methylsulfanyl-1-phenyl-1,3,5-triazin-2(1H)-one (6b).

This compound was prepared from $\mathbf{2 b}$ and phenyl isocyanate in $92 \%$ yield, $\mathrm{mp} 143^{\circ}$ (from diethyl ether); ir (potassium bromide): $v$ max 1705, 1684, 1571, 1479, $1286 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 2.14$ (s, $3 \mathrm{H}, \mathrm{CH}_{3}$ ), $2.56\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{SCH}_{3}\right)$, 7.21-7.56 (m, 5H, $\mathrm{C}_{6} \mathrm{H}_{5}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (deuteriochloroform): $\delta 14.2$ $\left(\mathrm{SCH}_{3}\right), 23.0\left(\mathrm{CH}_{3}\right), 127.1,129.7$ and 130.1 (5CHar), 136.3 (Car), $152.8\left(\mathrm{CCH}_{3}\right), 166.3(\mathrm{CO}), 183.3(\mathrm{SCN}) ; \mathrm{ms}: \mathrm{m} / \mathrm{z} 233$ $\left(\mathrm{M}^{+}\right), 218\left(\mathrm{M}^{+}-\mathrm{CH}_{3}\right), 118,77$.
Anal. Calcd. for $\mathrm{C}_{11} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{OS}: \mathrm{C}, 56.63$; $\mathrm{H}, 4.75$; N, 18.01. Found: C, 56.49; H, 4.78; N, 17.72.

4-Methylsulfanyl-1-phenyl-1,3,5-triazin-2(1H)-thione (7a).
This compound was prepared from 2a and phenyl isothiocyanate in $80 \%$ yield, $\mathrm{mp} 228^{\circ}$ (from diethyl ether); ir (potassium bromide): $v \max 1583,1442,1237 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 2.63\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{SCH}_{3}\right), 7.28-7.55\left(\mathrm{~m}, 5 \mathrm{H}, \mathrm{C}_{6} \mathrm{H}_{5}\right)$, 8.02 (s, $1 \mathrm{H}, \mathrm{NCH}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (hexadeuteriodimethylsulfoxide): $\delta$ $13.8\left(\mathrm{SCH}_{3}\right), 127.4,129.3$ and 129.4 (5CHar), 140.3 (Car), 156.2 (NCH), 177.6 (CS), 181.7 (SCN); ms: m/z $235\left(\mathrm{M}^{+}\right), 220\left(\mathrm{M}^{+}-\right.$ $\mathrm{CH}_{3}$ ), 104, 77.
Anal. Calcd. for $\mathrm{C}_{10} \mathrm{H}_{9} \mathrm{~N}_{3} \mathrm{~S}_{2}$ : C, 51.04; H, 3.85; N, 17.86. Found: C, 51.12; H, 4.03; N, 18.04.

6-Methyl-4-methylsulfanyl-1-phenyl-1,3,5-triazin-2(1H)-thione (7b).
This compound was prepared from $\mathbf{2 b}$ and phenyl isothiocyanate in $83 \%$ yield, $\mathrm{mp} 136-137^{\circ}$ (from diethyl ether); ir (potassium bromide): $v \max 1569,1558,1469,1450,1260 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 2.15\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CCH}_{3}\right), 2.61(\mathrm{~s}, 3 \mathrm{H}$, $\mathrm{SCH}_{3}$ ), 7.19-7.58 (m, 5H, C $\mathrm{C}_{6} \mathrm{H}_{5}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (deuteriochloroform): $\delta 14.3\left(\mathrm{SCH}_{3}\right), 24.3\left(\mathrm{CH}_{3}\right), 126.9,129.8$ and 130.5 ( 5 CHar ), 139.9 (Car), $164.7\left(\mathrm{CCH}_{3}\right), 177.5(\mathrm{CS}), 184.0(\mathrm{SCN}) ; \mathrm{ms}: \mathrm{m} / \mathrm{z}$ $249\left(\mathrm{M}^{+}\right), 234\left(\mathrm{M}^{+}-\mathrm{CH}_{3}\right), 202,118,77$.

Anal. Calcd. for $\mathrm{C}_{11} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{~S}_{2}$ : C, 52.99; H, 4.45; N, 16.85. Found: C, 52.77; H, 4.16; N, 16.63.

General Procedure for the Reaction Between Triazapentadienum Iodides 2 and Dimethyl Acetylenedicarboxylate.

Dimethyl acetylenedicarboxylate ( 2 mmoles ) was added to a solution of iodide 2 ( 2 mmoles) in dry methylene chloride $(10 \mathrm{ml})$ under a nitrogen atmosphere. The mixture was stirred
for 15 minutes at room temperature, cooled to $0^{\circ}$ then triethylamine ( 4 mmoles ) was added. The reaction mixture was stirred for 18 hours at room temperature. The solvent was removed and the residue was purified by flash chromatography (silica gel, methylene chloride/ethyl acetate 4:1). Products 8 and 9 were obtained.

Methyl 3-Methoxycarbonyl-7-dimethylamino-5-methylsulfanyl-4,6-diazaocta-2,4,6-trienoate (8).

This compound was prepared from $\mathbf{2 b}$ and dimethyl acetylenedicarboxylate in $79 \%$ yield (yellow oil, Rf methylene chloride $=0.2$ ); ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 2.17$ (s, $3 \mathrm{H}, \mathrm{CH}_{3}$ ), $2.40\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{SCH}_{3}\right), 2.96\left(\mathrm{~s}, 6 \mathrm{H}, \mathrm{N}\left(\mathrm{CH}_{3}\right)_{2}\right), 3.70$ and $3.79\left(2 \mathrm{~s}, 6 \mathrm{H}, 2 \mathrm{OCH}_{3}\right)$.

## 4,5-Dimethoxycarbonyl-2-methylsulfanylpyrimidine (9a).

This compound was prepared from 2a and dimethyl acetylenedicarboxylate in $48 \%$ yield, $\mathrm{mp} 133^{\circ}$ (from diethyl ether) [21].
4,5-Dimethoxycarbonyl-6-methyl-2-methylsulfanylpyrimidine (9b).

A solution of $\mathbf{8}(1.2 \mathrm{mmoles})$ in dry toluene $(40 \mathrm{ml})$ was heated for 18 hours at $110^{\circ}$. The solvent was removed and the residue was purified by flash chromatography (silica gel, methylene chloride/ethyl acetate $4: 1$ ) to give a yellow oil in $80 \%$ yield (Rf methylene chloride $=0.7$ ); ir (potassium bromide): $v$ max 1739, 1695, 1549, $1234 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta$ 2.61 and $2.62\left(2 \mathrm{~s}, 6 \mathrm{H}, \mathrm{CH}_{3}\right.$ and $\left.\mathrm{SCH}_{3}\right), 3.92$ and $3.96(2 \mathrm{~s}, 6 \mathrm{H}$, $2 \mathrm{OCH}_{3}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}$ (deuteriochloroform): $\delta 14.3\left(\mathrm{SCH}_{3}\right), 23.3$ $\left(\mathrm{CH}_{3}\right), 53.0$ and $53.4\left(2 \mathrm{OCH}_{3}\right), 119.7(\mathrm{CCO}), 155.4,165.0,166.2$ and $167.7\left(\mathrm{CCH}_{3}, \mathrm{NCCO}\right.$ and 2 CO$), 174.3(\mathrm{SCN}) ; \mathrm{ms}: \mathrm{m} / \mathrm{z} 256$ $\left(\mathrm{M}^{+}\right), 224\left(\mathrm{M}^{+}-\mathrm{CH}_{3} \mathrm{OH}\right), 196,166,138$.

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## REFERENCES AND NOTES

[1] D. L. Boger and S. M. Weinreb, Hetero Diels-Alder Methodology in Organic Synthesis, Academic Press, New York 1987, p 272.
[2] S. N. Mazumdar, I. Ibnusaud and M. P. Mahajan, Tetrahedron Lett., 27, 5875 (1986).
[3] P. Luthardt and E. U. Wurthwein, Tetrahedron Lett., 29, 921 (1988).
[4] J. Barluenga, M. Tomas, A. Ballesteros and A. L. Lopez, Tetrahedron Lett., 30, 4573 (1989).
[5] S. N. Mazumdar and M. P. Mahajan, Tetrahedron, 47, 1473 (1991).
[6] S. N. Mazumdar, S. Mukherjee, A. K. Sharma, D. Sengupta and M. P. Mahajan, Tetrahedron, 50, 7579 (1994).
[7] A. K. Sharma and M. P. Mahajan, Heterocycles, 40, 787 (1995).
[8] P. D. Dey, A. K. Sharma, S. N. Rai and M. P. Mahajan, Tetrahedron, 51, 7459 (1995).
[9] A. K. Sharma and M. P. Mahajan, Tetrahedron, 53, 13841 (1997).
[10] C. Friot, A. Reliquet, F. Reliquet and J. C. Meslin, Synthesis, 5, 695 (2000).
[11] A. Mai, M. Artico, G. Sbardella, S. Massa, A. G. Loi, E. Tramontano, P. Scano and P. La Colla, J. Med. Chem., 38, 3258 (1995).
[12] R. A. Nugent, S. T. Schlachter, M. J. Murphy, G. J. Cleek, T. J. Poel, D. G. Wishka, D. R. Graber, Y. Yagi, B. J. Keiser, R. A. Olmsted, L. A. Kopta, S. M. Swaney, S. M. Poppe, J. Morris, W. G. Tarpley and R. C. Thomas, J. Med. Chem., 41, 3793 (1998).
[13] C. O. Kappe and P. J. Roschger, J. Heterocyclic Chem., 26, 55 (1989) and references therein.
[14] C. O. Kappe, Tetrahedron , 49, 6937 (1993).
[15] S. N. Mazumdar, M. Sharma and M. P. Mahajan, Tetrahedron Lett., 28, 2641 (1987).
[16] P. de Tullio, B. Pirotte, F. Somers, S. Boverie, F. Lacan and J. Delarge, Tetrahedron, 54, 4935 (1998) and references therein.
[17] H. Bredereck, F. Effenberger and A. Hoffmann, Chem.Ber., 97, 61 (1964).
[18] C. W. Todd, J. H. Fletcher and D. S. Tarbell, J. Am. Chem. Soc., 65, 350 (1943).
[19] E. R. Garrett and D. J. Weber, J. Pharm. Sci., 59, 1383 (1970).
[20] Y. Watanabe, H. Usui, T. Shibano, T. Tanaka and M. Kanao, Chem. Pharm. Bull., 38, 2726 (1990).
[21] I. Ibnusaud, E. J. P. Malar and N. Sundaram, Tetrahedron Lett., 31, 7357 (1990).

